

#Jenny



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so many fake sites. this is the first one which worked! Many thanks

Name _____ Class _____ Date _____

Assessment

Chapter Test B

Chemistry of Life

Write the correct letter in the blank before each numbered term.

- | | |
|------------------------|--|
| _____ 1. covalent bond | a. involves transferred electrons |
| _____ 2. atom | b. loss of an electron |
| _____ 3. reduction | c. substance that cannot be broken down |
| _____ 4. compound | d. involves shared electrons |
| _____ 5. molecule | e. a substance composed of two or more elements |
| _____ 6. ionic bond | f. gain of an electron |
| _____ 7. oxidation | g. simplest part of an element |
| _____ 8. element | h. simplest part of a substance that can exist in a free state |

Write the letter of the most correct answer in the blank.

- _____ 9. Elements tend to undergo chemical reactions that _____ their stability.
a. increase c. do not change
b. decrease d. disrupt
- _____ 10. Sodium chloride (table salt) is an example of a compound formed by
a. hydrogen bonding. c. ionic bonding.
b. covalent bonding. d. electromagnetic bonding.
- _____ 11. When pure water dissociates, it forms equal numbers of
a. hydroxide ions and sodium ions.
b. hydronium ions and water ions.
c. hydroxide ions and water ions.
d. hydroxide ions and hydronium ions.
- _____ 12. Matter can change from a solid to a liquid by the addition of energy, which makes the particles in the matter
a. move faster. c. dissociate.
b. move slower. d. increase their specific heat.
- _____ 13. Charged particles that move around an atom's nucleus are
a. electrons. c. ions.
b. protons. d. neutrons.

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