

# Download File PDF Describing Motion Review And Reinforce Answers Pearson

#Jenny



Finally I get this ebook, thanks for all these I can get now!

#Rio



Cool! I'am really happy

#Markus Jensen



I did not think that this would work, my best friend showed me this website, and it does! I get my most wanted eBook

#Hun Tsu



wtf this great ebook for free?!

#Che Salsa



My friends are so mad that they do not know how I have all the high quality ebook which they do not!

#Diego Butler



so many fake sites. this is the first one which worked! Many thanks

## Chemistry: Chapter 1 & 2 Review

Name: \_\_\_\_\_

1. A mixture (is not) a chemical combination of substances.
2. In a compound the (atoms) (molecules) are (chemically) (physically) combined so that the elements that make up the compound (retain) (lose) their identities and (do) (do not) take on new set of properties.
3. The smallest identifiable unit of a compound is a(n) (MOLECULE) which is made up of (TWO OR MORE ATOMS), which are chemically bonded.
4. True or False? A mixture is always made up of a combination of elements. (FALSE) (TRUE) **COULD BE A COMBINATION OF COMPOUNDS**
5. In a mixture, the substances (lose) (retain) their identities.
6. In a mixture, the substances involved (can) (cannot) be separated by a simple physical process.
7. In a compound, the elements involved (can) (cannot) be separated by a simple physical process because the elements are (physically combined) (chemically bonded).
8. The smallest identifiable unit of an element is a(n) (ATOM).
9. Explain how to separate the sugar and water in a solution of sugar and water. (SOL. FOR EVAPORATE) (THE WATER GOES, LEAVING THE SUGAR BEHIND).
10. How would you separate a mixture of alcohol and water? (DISTILLATION - HEAT MIXTURE TO THE BOILING POINT (ALCOHOL) COLLECT VAPORS -> CONDENSE -> LIQUID)
11. How would you separate sand and water? (FILTER - WATER WOULD PASS THRU LEAVING THE SAND IN THE FILTER)
12. Classify the following as pure substances or as mixtures:

Gasoline	S / M	Sugar	S	Air	M
Mercury	S	Salt water	M	Oxygen	S
Dissolved Water	F	Blood	M	Mercury Oxide	S
13. What is the difference between extensive and intensive Properties?
14. Why are "intensive properties" so important?

[Download PDF version of :](#)  
**Describing Motion Review And Reinforce Answers Pearson**